

Oxidative Pyrolysis of Agricultural Residues in Gasification and Carbonization Processes

Xuan-Huynh Pham, Laurent van de Steene, Bruno Piriou, Sylvain Salvador

▶ To cite this version:

Xuan-Huynh Pham, Laurent van de Steene, Bruno Piriou, Sylvain Salvador. Oxidative Pyrolysis of Agricultural Residues in Gasification and Carbonization Processes. ICERE 2018 - 4th International Conference on Environment and Renewable Energy, Feb 2018, Da Nang, Vietnam. art.012032 - 8p., 10.1088/1755-1315/159/1/012032. hal-01919032

HAL Id: hal-01919032 https://imt-mines-albi.hal.science/hal-01919032

Submitted on 12 Nov 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

PAPER • OPEN ACCESS

Oxidative Pyrolysis of Agricultural Residues in Gasification and Carbonization Processes

To cite this article: Xuan-Huynh Pham et al 2018 IOP Conf. Ser.: Earth Environ. Sci. 159 012032

View the article online for updates and enhancements.

Related content

- Ammonium Removal Efficiency in Fixed Bed Reactor and Suspended Growth Reactor Using Ecotru Inoculation Rusyda Syahidah, Sudarno and Badrus Zaman
- Numerical Modelling of a Fast Pyrolysis Process in a Bubbling Fluidized Bed Reactor
 S Jalalifar, M Ghiji, R Abbassi et al.
- Modeling Lab-sized Anaerobic Fluidized Bed Reactor (AFBR) for Palm Oil Mill Effluent (POME) treatment: from Batch to Continuous Reactors Muhammad Mufti Azis, Hanifrahmawan Sudibyo and Wiratni Budhijanto



IOP ebooks™

Bringing you innovative digital publishing with leading voices to create your essential collection of books in STEM research.

Start exploring the collection - download the first chapter of every title for free.

doi:10.1088/1755-1315/159/1/012032

Oxidative Pyrolysis of Agricultural Residues in Gasification and Carbonization Processes

Xuan-Huynh Pham^{1,2}, Laurent Van De Steene^{1,2}, Bruno Piriou¹, Sylvain Salvador³

Email: huynh.pham_xuan@cirad.fr

Abstract. Oxidative pyrolysis, in which heat is provided by the partial oxidation, occurs in many fixed bed reactors of air staged gasification and advanced carbonization processes. In these reactors, an Oxidation Zone (OZ) propagates and separates the virgin biomass from the produced char, and self-sustains the process. To investigate the energy transfer efficiency and to characterize the OZ features, oxidative pyrolysis was performed in a 20 cm diameter fixed bed reactor with Wheat straw pellets. Temperatures and biomass bed height were measured continuously during the experiment. The OZ propagated at 0.41 cm/min and consumed about 11 % of the biomass. The role of bed compaction on the effective propagation velocity was highlighted. Char yield was measured to 28.6 % and char accounted for 39.56 % of the total energy content of the Wheat straw. Maximum temperature in the bed reached 780 °C. This paper proposes a new insight into the OZ features when it propagates in a biomass fixed bed.

1. Introduction

Agricultural residues must be disposed in order to avoid contamination, pest growth, occupying large expanses of land and hindering agricultural work. In addition, agricultural residues can be exploited for recovering specific components or useful energy. Therefore, valorization of agricultural residues for energy purposes has multiple advantages: minimizing the environmental impacts, increasing economic benefits for the farmers, and reducing the reliance on fossil fuels [1], [2].

Holding a huge potential and the carbon-rich composition, agricultural residues are being considered as the major contributors of resources for energy and material productions. The global annual production of agricultural residues of the six most important crops (barley, maize, rice, soybean, sugar cane and wheat) in 227 countries and territories of the world is reported to be 3.7×10^{12} kg of dried matter. This residues production is proportional to crop yields [1], [3].

Thermo-chemical conversion, i.e. combustion, gasification and pyrolysis can convert agricultural residues into variety of useful energy types such as heat, electricity and materials like charcoal, syngas and bio-oil. Pyrolysis for charcoal production, carbonization, is one of the most widely used technologies. It is normally known as the thermal decomposition of organic components of biomass in an inert gas atmosphere, at moderate temperature. In such technology, the external energy source is needed to provide heat for the pyrolysis reactions. However, some current studies have shown that the partial oxidation of biomass and products, so called *oxidative pyrolysis*, is an effective way to provide

¹CIRAD, UPR BioWooEB, F-34398 Montpellier, France.

² Department of Energy, University of Science and Technology of Hanoi, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Hanoi, Vietnam.

³ RAPSODEE, CNRS UMR 5203, Mines-Albi, Campus Jarlard, 81013 Albi Cedex 09, France.

Content from this work may be used under the terms of the Creative Commons Attribution 3.0 licence. Any further distribution of this work must maintain attribution to the author(s) and the title of the work, journal citation and DOI.

doi:10.1088/1755-1315/159/1/012032

energy to the process [4], [5]. Oxidative pyrolysis allows the autothermal operation; and contributes to reduce complexity of the system and operating cost.

Here, we characterized the oxidation features in terms of propagation velocity, temperature, char yield, equivalence ratio and energy biomass conversion to pyrolysis products.

2. Material and equipment

2.1. Material

In this study, we used Wheat straw as a specific agricultural residue (Table 1). The pellet form was used to avoid the bridging problem occurring in stratified fixed bed reactors [6]–[8] and to ensure the homogeneity in the biomass bed properties.

Density (kg/m³)	Moisture (wt%)	Proximate analysis (wt%, d.b.)			Ultimate analysis (wt%, d.a.f.)				LHV,
		Volatile mater	Ash	Fixed carbon	C	Н	N	О	(MJ/kg)
549	8.75	78.75	8.17	13.08	58.91	5.07	1.10	34.92	17.45
± 1.42	±0.05	±1.89	±0.09	±1.89	±0.18	±0.04	±0.10	±0.24	±0.12

Table 1. Wheat straw pellets properties

2.2. Equipment description

Oxidative pyrolysis was studied in a fixed bed reactor (Figure 1). It consists of a 310 type refractory steel tube (20cm I.D.; 160 cm L) surrounded by a 30 cm thick refractory wool insulation. Thermocouples are installed every 10 cm inside the bed, allowing continuous measurement of temperature. In addition, a laser beam device is placed above the biomass bed to continuously measure the bed height during experiments allowing the calculation of the bed compaction.

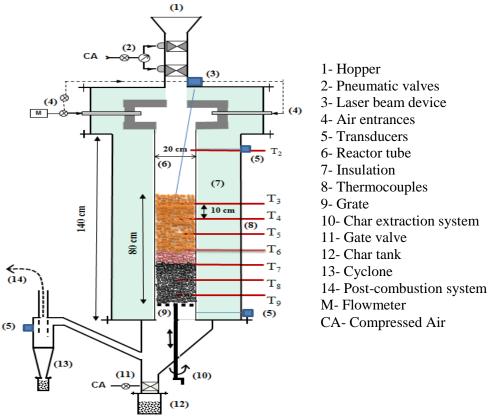


Figure 1. The fixed bed reactor

doi:10.1088/1755-1315/159/1/012032

The experimental procedure consists in igniting the OZ at the bottom of a 70 cm high biomass bed and following its upward propagation (about 10 cm of charcoal was used for the ignition). For this purpose, air flow is supplied at the top of the reactor and passes down through the biomass bed. As an example, evolution of temperatures at the different thermocouples presented in figure 2a shows the propagation of the OZ through a Wheat straw pellets bed while figure 2b shows the bed height reduction during the test.

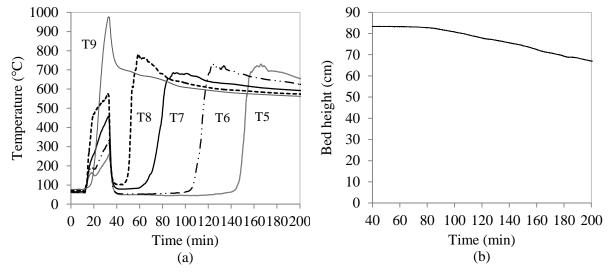


Figure 2. Typical experimental results with an air flux of 0.022 kg/m²s:

(a) Temperature profile and (b) Bed height reduction

In Figure 2a, first the temperature in T9 reach 950-1000 $^{\circ}$ C after about 30 mins following the ignition. Then, the air flux was reduced to the studied value of 0.022 kg/m²s involving a decrease of the temperature tends to about 600 $^{\circ}$ C up to the end of the experiment. Then, the oxidation zone propagates upward from T9 position to the top of the bed, in T5 position. The passage is observed by the increase of each temperature measured by thermocouples located all along the biomass/char bed each 10 cm. Thus, the temperature profiles versus time can be used for determination of the OZ propagation velocity in the biomass bed.

The bed height remains constant up to 80 mins, then, continuously reduces up to the end of the test (Figure 2b). This fact reveals that the oxidation zone propagates in a stationary regime only after reaching T7. Thus, in further discussion, we focus on the zone behavior from T7 to T5, i.e. 25cm to 45 cm respectively from the bottom of the bed. The bed height measurement allows the determination of the bed compaction velocity, and thus the biomass consumption rate.

3. Method for characterizing the oxidation zone features

From these temperature and bed height measurements, a new method has been proposed to calculate the upward propagation velocity of the oxidation zone. In this reverse smoldering configuration, the OZ propagates upward while the bed compaction makes biomass going down into the oxidation zone. The bed compaction velocity needs to be taken into account when determining the biomass consumption rate (Figure 3).

doi:10.1088/1755-1315/159/1/012032

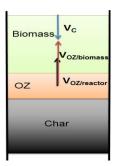


Figure 3. Simplified sketch illustrating OZ velocities and compaction velocity in fixed bed

We define $V_{OZ/reactor}$ (cm/min) as the apparent upward velocity. It is the OZ velocity related to the reactor, the one we observe from outside of the reactor. It is measured by the temperature evolution inside the reactor as described in next section. $V_{OZ/biomass}$ (cm/min) is defined as the velocity of the oxidation zone related to biomass, which is an effective velocity of propagation of the oxidation zone in the biomass. V_C (cm/min) is defined as the compaction velocity; it is measured by the laser during experiments and allows the calculation of Voz/biomass by following equation:

$$V_{OZ/biomass} = V_{OZ/reactor} + V_C \tag{1}$$

Here, we assume that the bed compaction only occurs within the oxidation zone. In other words, there is no compaction in the bed of biomass and in the bed of produced char.

3.1. Measurement of V_{OZ/reactor}

The position of the oxidation zone at the elevation of a given thermocouple Ti is indicated when dTi/dt is maximum. We define t_{OZ}^i as the time coresponding to the passage of the OZ at that elevation (Figure 4).

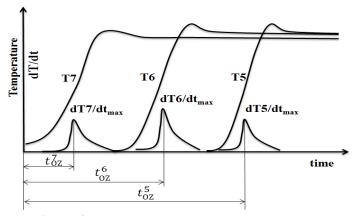


Figure 4. OZ passage time determination.

The position of the oxidation zone in the reactor versus time can be plotted (Figure 5). The upward propagation velocity of the OZ can be determined from the derivative of the position function. For this purpose, a fitting of the experimental data was performed (Figure 6).

doi:10.1088/1755-1315/159/1/012032

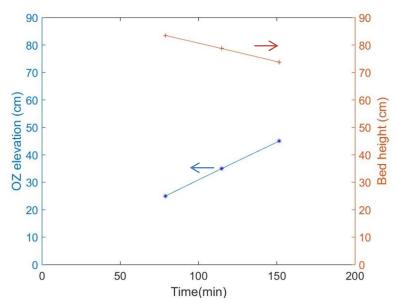


Figure 5. The evolution of the OZ propagation rate along a Wheat straw bed (air flux: 0.022 kg/m²s).

As shown in Figure 5, the oxidation zone reached T7 position (Z=25cm) at around 80 mins, then propagated upward and reached T6 (Z=35cm) at 118 mins and T5 (Z=45cm) at 156 mins. The total bed height reduced from 82 cm when OZ arrived at T7 (25 cm) to around 74 cm when OZ arrived at T5 (45 cm).

3.2. Measurement of the biomass consumption and char production

The hypothesis that the compaction happens only within the oxidation zone allows the estimation of the apparent OZ char velocity ($V_{oz/char}$):

$$V_{OZ/char} = V_{OZ/reactor}$$
 (2)

The biomass consumption rate then is calculated as following equation:

$$\dot{m}_{biomass} = V_{OZ/biomass} * A_{reactor} * \rho_{biomass}$$
 (3)

Char production rate:

$$\dot{m}_{char} = V_{OZ/reactor} * A_{reactor} * \rho_{char} \tag{4}$$

 ρ_{char} is the density of char which was measured about 236 kg/m³.

3.3. Equivalence ratio (ER)

The ratio of air to biomass is written:

$$Air: biomass = \frac{Air\ mass\ flux}{Biomass\ consumption\ rate}$$
(5)

Thus, the ER is classically defined as follow:

$$ER = \frac{Air: biomass}{Air: biomass_{stoichiometry}}$$
 (6)

Note that the Air: biomass stoichiometry is 5.38 calculated from the ultimate analysis (Table 1) of Wheat straw pellets.

4. Results and discussions

4.1. Propagation velocities of the oxidation zone

doi:10.1088/1755-1315/159/1/012032

Propagation velocities include the apparent velocity ($V_{OZ/reactor}$) and the effective velocity ($V_{OZ/biomass}$). In figure 6, the apparent velocity of the oxidation zone was shown about 0.28 cm/min while the bed compaction velocity was found to be 0.13 cm/min. As a consequence, the effective velocity of the oxidation zone in the biomass bed was calculated to 0.41 cm/min. Bed compaction and propagation velocities were almost constant during the test period, which confirm that a stationary regime was reached in the zone from 25 cm to 45 cm in the reactor (from T7 and T5).

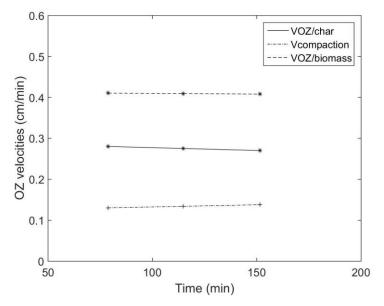


Figure 6. OZ propagation velocities with an air flux of 0.022 kg/m²s.

These results show that in downdraft fixed bed configuration, the bed compaction accelerated the propagation of the oxidation zone. However, few studies [9], [10] have taken the bed compaction into account when calculating the propagation velocity of the oxidation zone.

4.2. Temperature

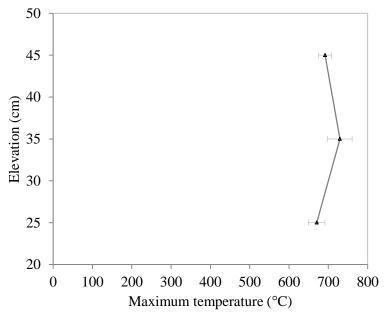


Figure 7. Maximum temperature profile with an air flux of 0.022 kg/m²s.

doi:10.1088/1755-1315/159/1/012032

The maximum temperature (Figure 7) in the bed of Wheat straw pellets oscillated between $650 \,^{\circ}\mathrm{C}$ and $780 \,^{\circ}\mathrm{C}$ and no hot spot temperature along the bed was observed. Temperature is a major influencing factor regarding pyrolysis products, yields and composition. It is reported that pyrolysis temperature is increased, the char yield decreased [11], [12].

4.3. Biomass consumption rate, char production rate and equivalence ratio

Propagation velocities allow the calculation of the biomass consumption rate and the char production rate (Figure 8). In our operating conditions, biomass consumption rate was calculated to 4.2 kg/h and the char production rate to 1.2 kg/h. The char yield defined as the ratio of char production rate over biomass consumption rate was 28.6 %. This result shows that char yield in the oxidative pyrolysis was in line with the typical slow pyrolysis processes [13].

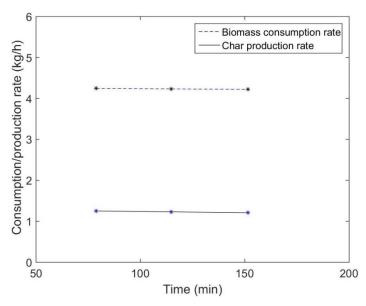


Figure 8. Biomass consumption rate and char production rate with an air flux of 0.022 kg/m²s.

The energy transferred from biomass to the char was calculated from the measurements of char yield and the lower heating value (LHV) of charcoal and biomass.

The LHV was measured to 24.14 MJ/kg and 17.45 MJ/kg for charcoal and biomass, respectively.

$$E_E = \frac{\dot{m}_{char} * LHV_{char}}{\dot{m}_{biomass} * LHV_{biomass}} = 39.56 \%$$
 (7)

Energy transfer to the char was about 39.56 %. The rest of energy in biomass was transferred to the other products such as permanent gases and liquid fuels.

Finally, the ER of our autothermal pyrolysis process was calculated to be 0.11. It means that the oxidation zone consumed 11 % biomass to provide heat for self-sustain the process.

5. Conclusions

The aim of this work was to investigate the oxidative autothermal pyrolysis in fixed beds of agricultural residues pellets. The OZ features in terms of propagation velocity, temperature, char yield and equivalence ratio were discussed. Regarding the air mass flux of $0.022 \text{ kg/m}^2\text{s}$, the oxidation zone propagates at an effective velocity of 0.41 cm/min in the biomass bed. The maximum temperature oscillated between 650°C and 780°C. The OZ consumes 11 % of the Wheat straw to provide energy for the autothermal process. Char yield and energy transferred from biomass to the char were measured to 28.6 % and 39.56 %, respectively.

These results suggest the direction for the valorisation agricultural residues for energy purpose by oxidative pyrolysis. The future work will focus on the characterization the oxidation zone features in fixed bed of various biomass types.

doi:10.1088/1755-1315/159/1/012032

References

- [1] Bentsen N S, Felby C and Thorsen B J 2014 Agricultural residue production and potentials for energy and materials services *Prog. Energy Combust. Sci.* **40** 59–73
- [2] Malico I, Carrajola J, Gomes C P and Lima J C 2016 Biomass residues for energy production and habitat preservation. Case study in a montado area in Southwestern Europe *J. Clean. Prod.* **112** 3676–83
- [3] Colantoni A, Evic N, Lord R, Retschitzegger S, Proto A R, Gallucci F and Monarca D 2016 Characterization of biochars produced from pyrolysis of pelletized agricultural residues *Renew. Sustain. Energy Rev.* **64** 187–94
- [4] Anca-Couce A, Zobel N, Berger A and Behrendt F 2012 Smouldering of pine wood: Kinetics and reaction heats *Combust. Flame* **159** 1708–19
- [5] Daouk E, Van de Steene L, Paviet F, Martin E, Valette J and Salvador S 2017 Oxidative pyrolysis of wood chips and of wood pellets in a downdraft continuous fixed bed reactor *Fuel* **196** 408–18
- [6] Ummadisingu A, Sheth P and Babu B V 2010 Experimental Studies on Gasification of Pine Wood Shavings in a Downdraft Biomass Gasifier **31** 2–7
- [7] Paulauskas R, Striūgas N and Misiulis E 2014 Experimental and theoretical investigation of wood pellet shrinkage during pyrolysis 1–11
- [8] Guo F, Dong Y, Dong L and Guo C 2014 Effect of design and operating parameters on the gasification process of biomass in a downdraft fixed bed: An experimental study *Int. J. Hydrogen Energy* **39** 5625–33
- [9] Pillier L, Idir M, Molet J, Matynia A and De Persis S 2015 Experimental study and modelling of NOx formation in high pressure counter-flow premixed CH4/air flames *Fuel* **150** 394–407
- [10] Mahapatra S, Kumar S and Dasappa S 2016 Gasification of wood particles in a co-current packed bed: Experiments and model analysis *Fuel Process. Technol.* **145** 76–89
- [11] Mahapatra S and Dasappa S 2014 Experiments and analysis of propagation front under gasification regimes in a packed bed *Fuel Process. Technol.* **121** 83–90
- [12] Kumar M, Gupta R C and Sharma T 1992 Effects of carbonisation conditions on the yield and chemical composition of Acacia and Eucalyptus wood chars *Biomass and Bioenergy* **3** 411–7
- [13] Katyal S, Thambimuthu K and Valix M 2003 Carbonisation of bagasse in a fixed bed reactor: Influence of process variables on char yield and characteristics *Renew. Energy* **28** 713–25

Acknowledgements

Authors would like to thank Mr. Jérémy Valette for his technical assistance and Dr. Jean-Michel Commandréfor his advices and material supports for this study.

This work is funded by the Vietnamese International Education Development (VIED) and by the agreement with the CIRAD-France.